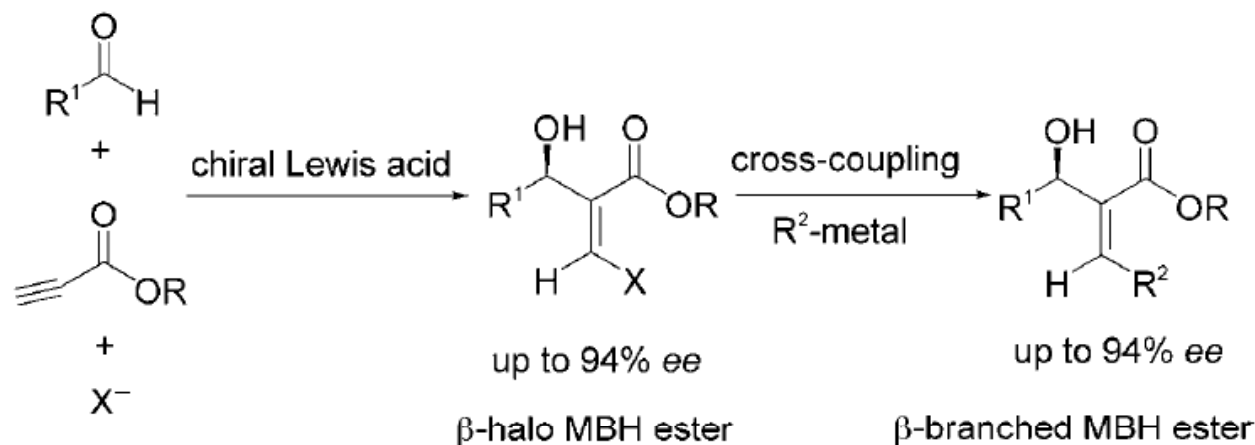


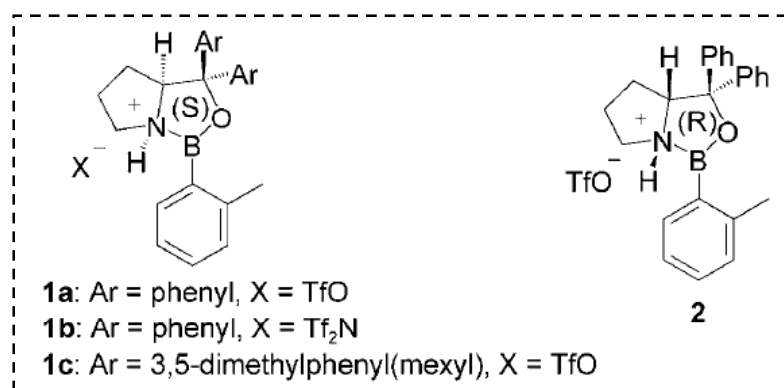
Enantioselective Synthesis of β -Iodo Morita–Baylis–Hillman Esters by a Catalytic Asymmetric Three-Component Coupling Reaction**

Bidyut Kumar Senapati, Geum-Sook Hwang, Sungil Lee, and Do Hyun Ryu*

Angewandte Chemie Int. Ed. **2009**, early view.



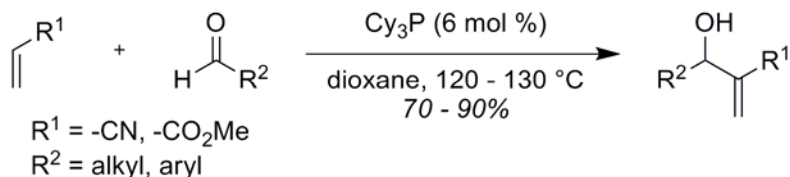
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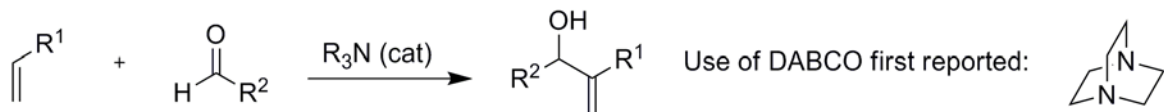
Nolan D. Griggs - 5-22-2009

The Morita-Baylis Hillman Reaction - Overview

- First example by Morita and co-workers in 1968:
Morita, K., Suzuki, Z., Hirose, H. *Bull. Chem. Soc. Jpn.* **1968**, 41, 2815.



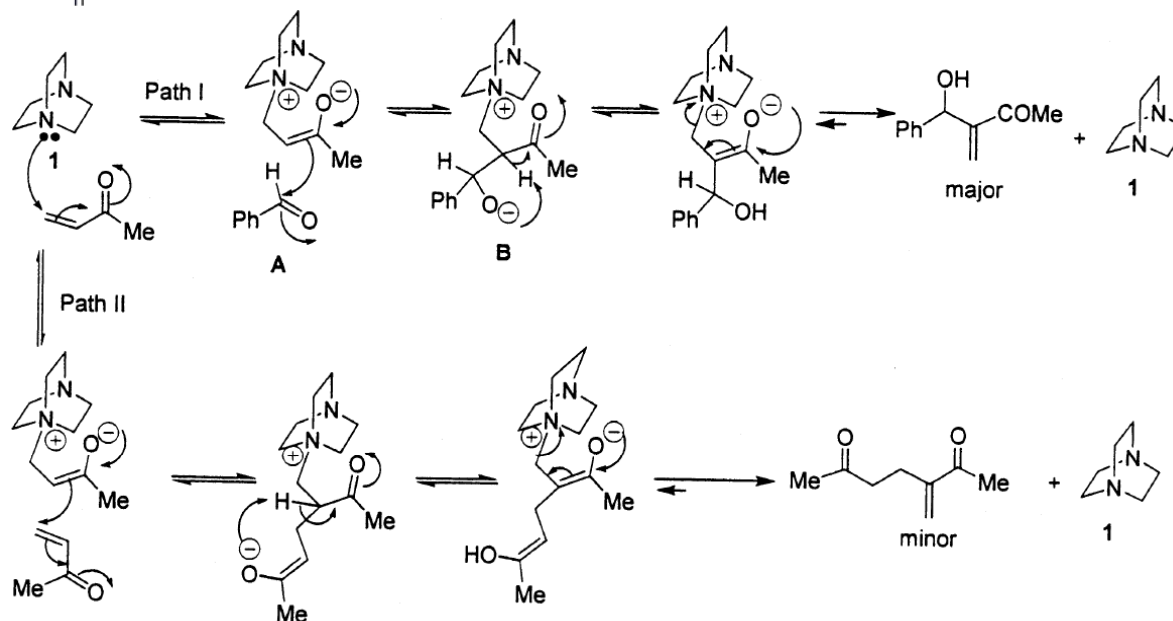
- Later published 1972 by Baylis and Hillman in a German patent:
Baylis, A.B., Hillman, M.E.D. *Acrylic Compounds*. De 2155113, **1972** (Celanese Corp.).



- Proposed Mechanism:

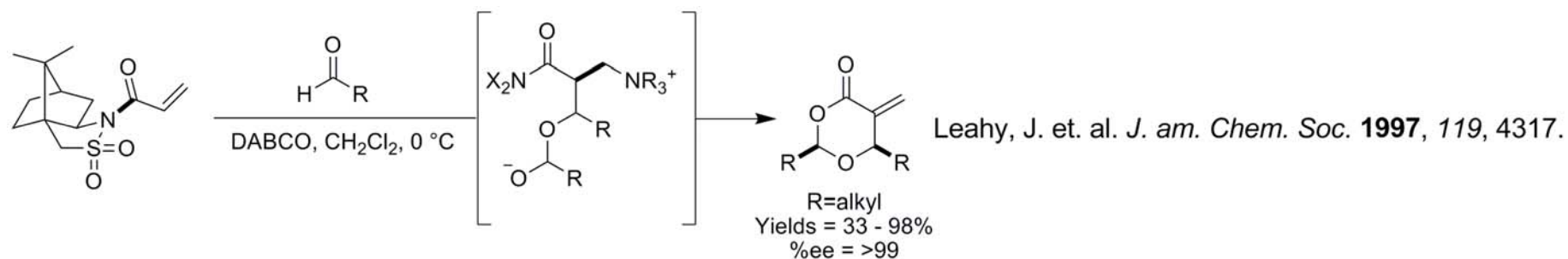
- Reviews:

- *ACIEE* **2007**, 46, 4614
- *Eur. J. Org. Chem.*, **2007**, 2905
- *Chem. Soc. Rev.*, **2007**, 36, 1581
- *Chem. Rev.* **2003**, 103, 811.

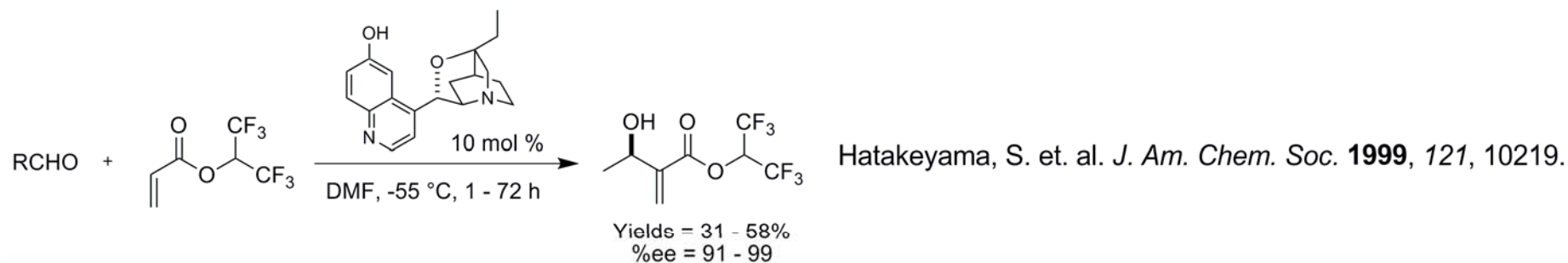
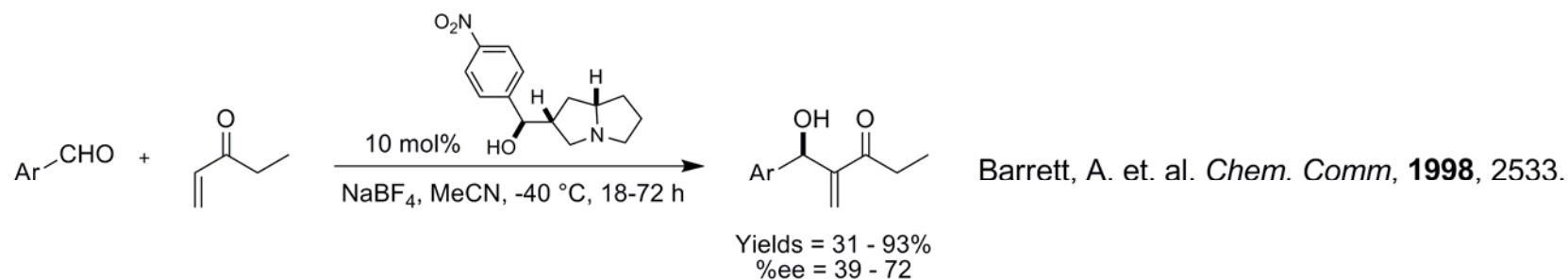


Asymmetric Morita-Baylis-Hillman Reaction

- Auxillary based:



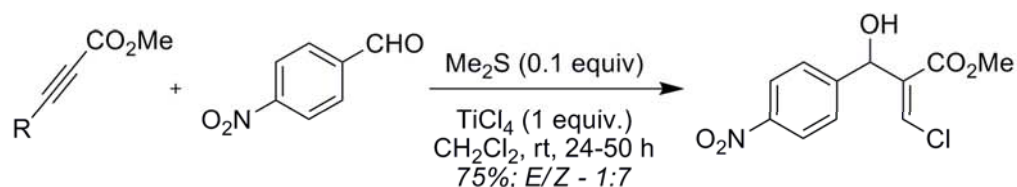
- Organocatalytic methods:



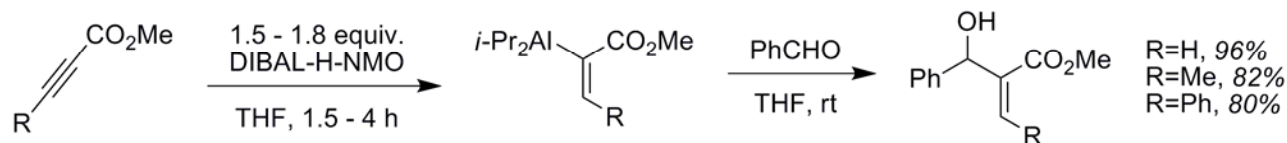
For a review, see: Langer, P. *Angew. Chemie. Int. Ed.*, **2000**, 39, 3049.

Stereochemical Control of β -branched MBH products

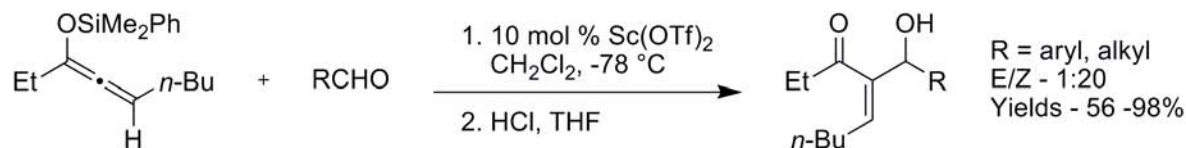
- Kataoka and co-workers - *Angew. Chemie. Int. Ed.* **2000**, 39, 2358.



- Ramachandran and co-workers - *J. Org. Chem.* **2003**, 68, 9310.

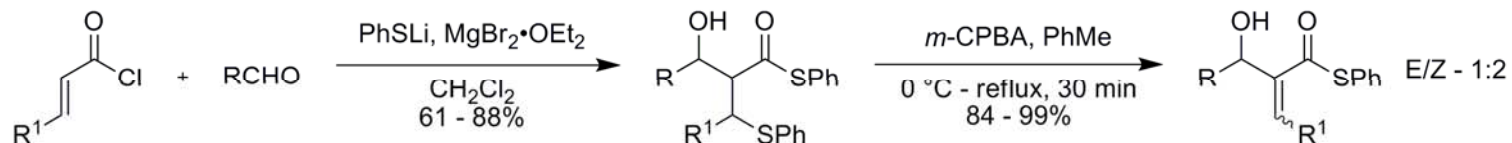


- Scheidt and co-workers - *J. Am. Chem. Soc.* **2006**, 128, 15382.



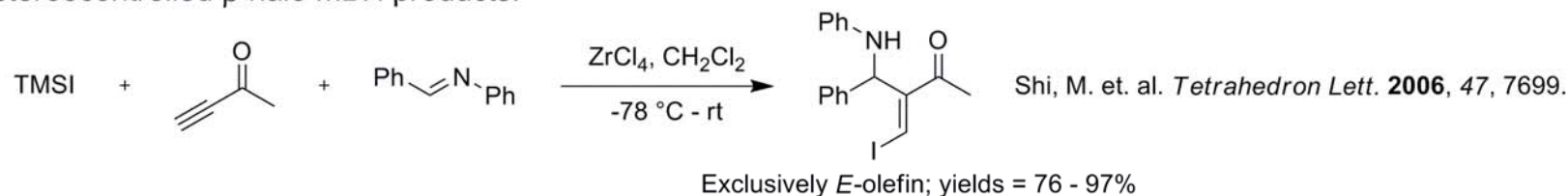
- For an asymmetric version using $(\text{Salen})\text{CrSbF}_6$ w/ ee = 61-94%, see: *Angew. Chemie. Int. Ed.*, **2007**, 46, 7806.

- Coltart and co-workers - *Org. Lett.*, **2008**, 10, 4819.

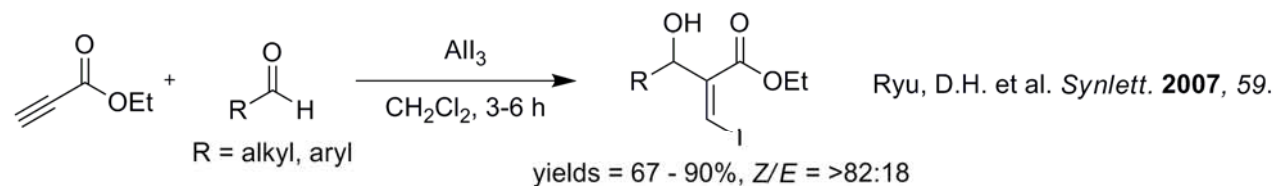
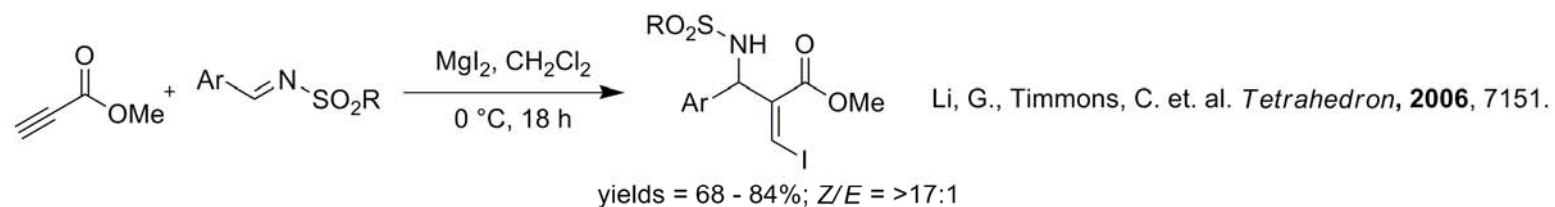


Synthesis of β -halo MBH products

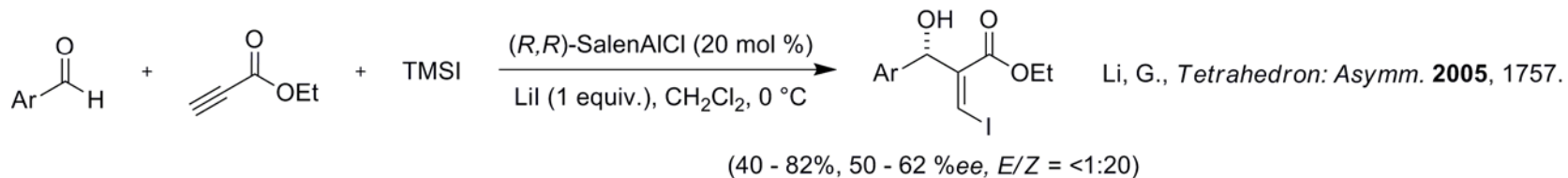
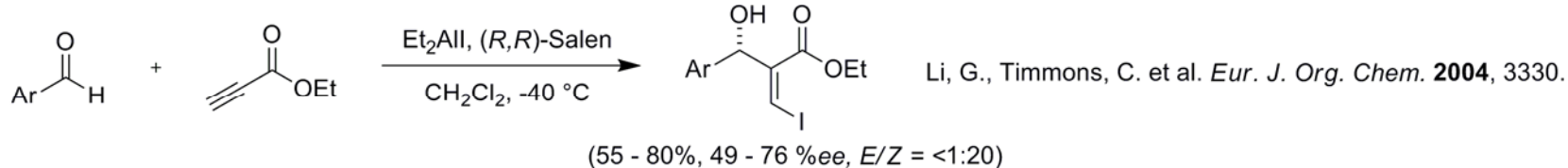
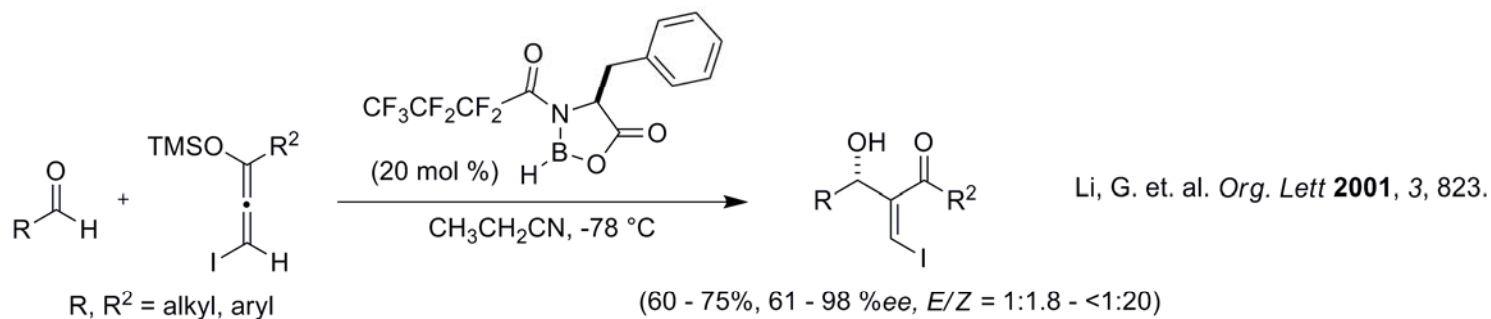
- *E*-stereocontrolled β -halo MBH products:



- *Z*-stereocontrolled β -halo MBH products: Review - *Recent Res. Dev. Org. Bioorg. Chem.* **2001**, 49.

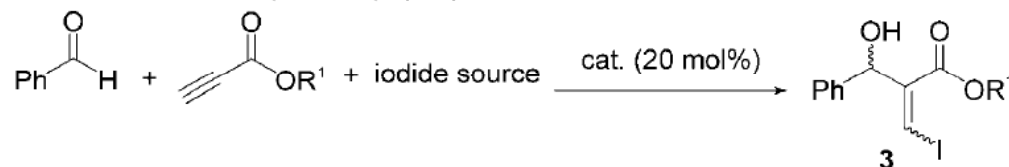


Asymmetric Synthesis of β -iodo MBH products

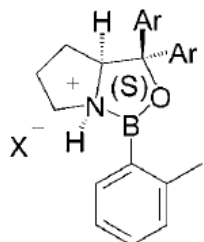


Title Paper: Catalytic Asymmetric Synthesis of β -Iodo MBH Products

• Initial Screening:



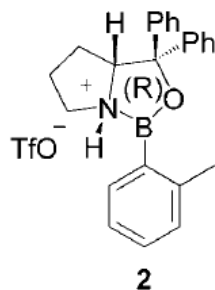
Entry	Cat.	R ¹	Reaction conditions	Yield [%] ^[b]	Z/E ^[c]	ee [%] ^[d]
1	1a	Et	<i>n</i> Bu ₄ NI, CH ₂ Cl ₂ , -40 °C, 10 h	26	85:15	20
2	1a	Et	TMSI, CH ₂ Cl ₂ , -40 °C, 3 h	68	90:10	77
3	1a	Et	TMSI, CH ₃ CH ₂ CN, -40 °C, 10 h	38	88:12	69
4	1a	Et	TMSI, toluene, -78 °C, 5 h	85	92:8	84
5	1b	Et	TMSI, CH ₂ Cl ₂ , -78 °C, 3 h	92	94:6	67
6	1a	Me	TMSI, CH ₂ Cl ₂ , -78 °C, 1.5 h	92	99:1	84
7	1a	<i>t</i> Bu	TMSI, CH ₂ Cl ₂ , -78 °C, 10 h	0	—	—
8	1a	Et	TMSI, CH ₂ Cl ₂ , -78 °C, 2 h	93	>99:1	87 ^[e]
9	2	Et	TMSI, CH ₂ Cl ₂ , -78 °C, 2 h	90	>99:1	87 ^[f]
10	1c	Et	TMSI, CH ₂ Cl ₂ , -78 °C, 1 h	95	>99:1	94 ^[e]



1a: Ar = phenyl, X = TfO

1b: Ar = phenyl, X = Tf₂N

1c: Ar = 3,5-dimethylphenyl(mexyl), X = TfO

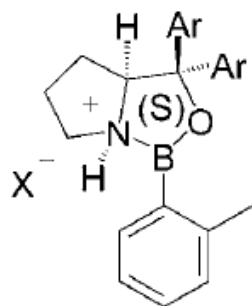


2

[a] Reactions run with 1.0 mmol of benzaldehyde, 2.0 mmol of ethylpropiolate, 1.5 mmol of the iodide source, and 0.2 mmol of catalyst. [b] Yield of isolated product. [c] Determined after separation by column chromatography. [d] Determined by HPLC on a chiral stationary phase. [e] The absolute configuration of **3** was determined to be *R* enriched. For details see the Supporting Information. [f] The absolute configuration of **3** was determined to be *S* enriched.

Title Paper: Catalytic Asymmetric Synthesis of β -Iodo MBH Products

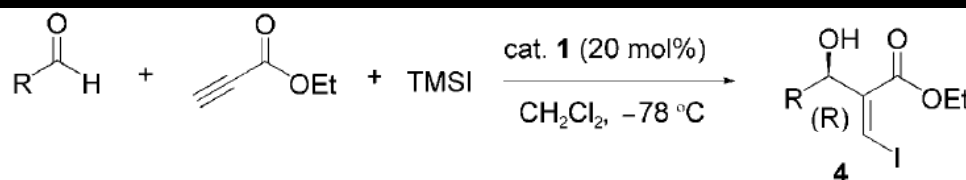
• Scope and Limitations:



1a: Ar = phenyl, X = TfO

1b: Ar = phenyl, X = Tf₂N

1c: Ar = 3,5-dimethylphenyl(mexyl), X = TfO

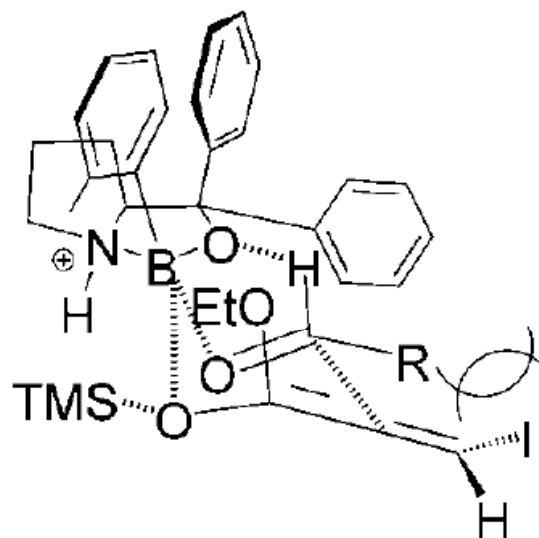
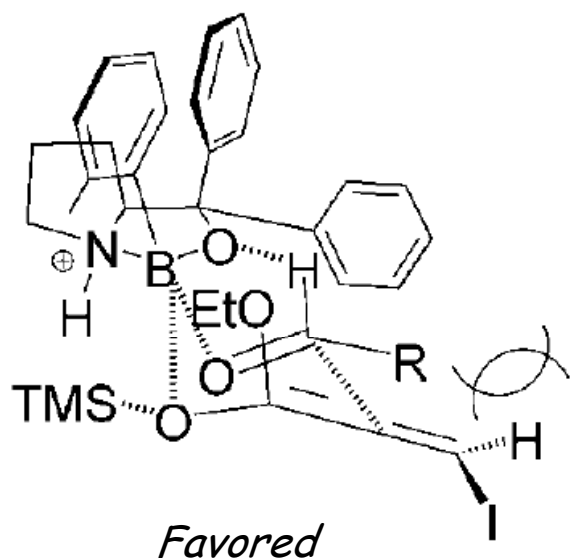


Entry	Catalyst	R	t [h]	Yield [%] ^[b]	Z/E ^[c]	ee [%] ^[d]
1	1c	Ph	1	95	> 99:1	94
2	1c	4-F ₆ H ₄	3	92	> 99:1	92
3	1a	4-CF ₃ C ₆ H ₄	4	75	99:1	92
4	1a	4-ClC ₆ H ₄	2	99	99:1	96
5	1a	2-BrC ₆ H ₄	5	95	> 99:1	90
6	1a	4-BrC ₆ H ₄	5	90	96:4	93
7	1a	4-CNC ₆ H ₄	6	91	97:3	95
8	1a	4-NO ₂ C ₆ H ₄	30	66	92:8	90
9	1c	4-MeC ₆ H ₄	1.5	92	99:1	62
10	1c	4-PhC ₆ H ₄	1.5	95	92:8	90
11	1a	2-naphthyl	12	65	98:2	91
12 ^[e]	1b	<i>n</i> Pr	8	72	96:4	93
13 ^[e]	1b	<i>n</i> -hexyl	12	61	97:3	90
14 ^[e]	1b	<i>i</i> Pr	12	50	95:5	90

[a] Reactions run with 1.0 mmol of aldehyde, 2.0 mmol of ethylpropiolate, 1.5 mmol of TMSI, and 0.2 mmol of catalyst. [b] Yield of isolated product. [c] Determined after separation by column chromatography. [d] Determined by HPLC on a chiral stationary phase. [e] Reaction run using 2.5 equivalents of ethyl propiolate and 2.0 equivalents of TMSI at -60 °C.

Title Paper: Catalytic Asymmetric Synthesis of β -Iodo MBH Products

- Proposed Transition-state model:



For similar oxazaborolidinium catalyst complexes involving aldehydes, see:

Ryu, D.H., Corey, E.J. *J. Am. Chem. Soc.* **2003**, *125*, 6388.

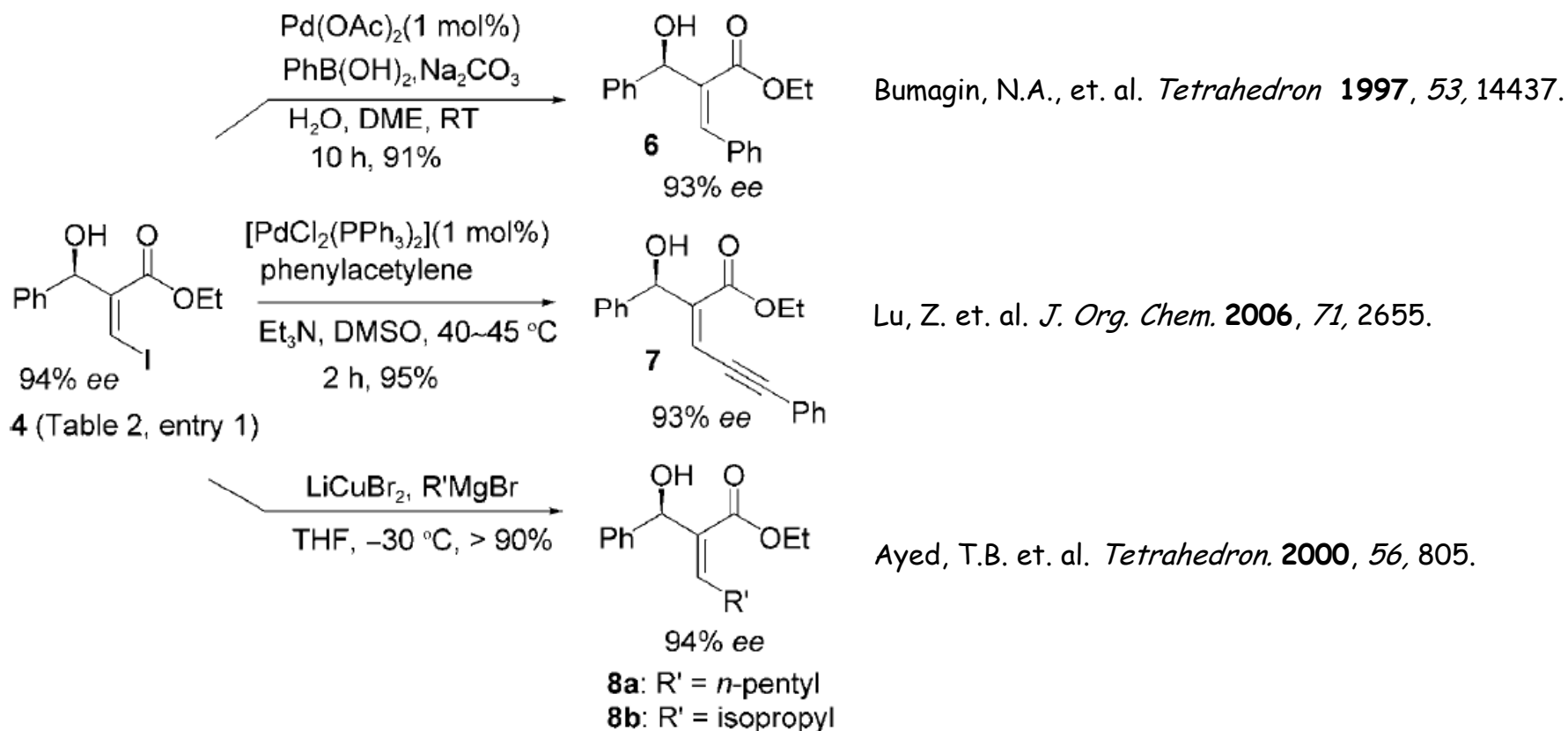
Ryu, D.H., Corey, E.J. *J. Am. Chem. Soc.* **2004**, *126*, 8106.

For discussion of pentacoordinated boron species, see:

Maruoka, J. et al. *J. Am. Chem. Soc.* **1998**, *120*, 5327.

Title Paper: Catalytic Asymmetric Synthesis of β -Iodo MBH Products

- Further Application of MBH adducts:



Conclusions

- Ryu, D.H. and co-workers have developed a catalytic, highly enantioselective three-component coupling between an aldehyde, ethyl propiolate, and TMSI to give optically active (*Z*)- β -iodo Morita-Baylis-Hillman esters.
- Both enantiomers are accessible using the appropriate, readily available oxazaborolidinium catalyst, and the absolute configuration matches that predicted by the transition state model.
- Further functionalization of the product vinyl iodides led to highly functionalized, optically active *Z*-olefin-containing products.